

Oxidation of Sunflower Oil Using *Annona Squamosa* and *Vangueria Madagascariensis* Methanolic Extracts

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Submitted: 2024, Nov 29; Accepted: 2024, Dec 23; Published: 2025, Jan 07

Citation: Fadul, S. E. K. M., Ahmed, Y. M., Mariod, A. A. (2025). Oxidation of Sunflower Oil Using *Annona Squamosa* and *Vangueria Madagascariensis* Methanolic Extracts. *Adv Nutr Food Sci*, 10(1), 01-08.

Abstract

The purpose of this study was to assess how *Annona squamosa* Gishta and *Vangueria madagascariensis* Kikir methanolic extract affected sunflower oil stabil-ity. Measurements of the formation of peroxide value PVs and the Conjugated Diene CD Test were used to assess the impact of *Annona squamosa*(Gishta) and *Vangueria madagascariensis* (Kirkir) Methanolic extracts (Leaves, seed cake and bark) on the ox-idative stability of sunflower oil stored in the dark at 70°C for 72 hours, in compari-son to the control and BHT(butylated hydroxytoluene). The best oxidative stability was demonstrated by plant extracts of Gishta and Kirkir leaves (500 mg), which reduced PVs and decreased C D. These extracts also successfully prevented the formation of peroxides in sunflower oil when compared to 20 mg of BHT.

Keywords: *Annona Squamosa*, *Vangueria Madagascariensis*, Oxidative Stability, Conjugated Diene

1. Introduction

The ability of edible oils to withstand oxidation during handling, processing, and stor- age is known as oxidative stability. Lipid autoxidation is a spontaneous reaction that takes place between unsaturated fatty acids and molecular oxygen via a chain reaction of free radicals that includes the production of hydroperoxides, fat free radicals, and peroxide free radicals [1]. Sunflower oil's high PUFA content makes it extremely prone to lipid oxidation [2]. The analysis of primary oxidation compounds is always performed using the peroxide value (PV) or the conjugated dienes content (CD) since polyunsaturated fatty acid hy- dro peroxides, which are the most susceptible to oxidation, exhibit a strong absorbance at 232 nm [3].

One of the most often determined quality indicators in the production, storage, and sell- ing of oil is Peroxide Value (PV). PV quantifies the total number of peroxides pro- duced as a byproduct of primary oil oxidation and indicates the degree of oxidation in the material [4]. Hydroperoxides, often known as peroxides, are the main byproducts of lipid peroxida- tion. According to [5], the peroxide value estimation results so clearly show lipid autoxidation, evaluating the generation of peroxide value at 70°C, the ef- fects of different plant extracts on the oxidative stability of sunflower oil— which is particularly rich in linoleic fatty acid—were assessed.

Also, the concentrations of con- jugated diene hydroperoxides and conjugated diene non-hydroperoxides as the primary and secondary, respectively, oxidation products are represented by the conjugated diene value (CDV). The value can be ascertained with great ease and sensitivity [6]. Leaves, seed cakes and bark extracts from *Annona squamosa* and *Vangueria madagascariensis* are rich in polyphenols and flavonoids and show antioxidant properties [7,8]. The aim of this study is to assess the impact of extracts from *Annona squamosa* and *Vangueria mada- gascariensis* on the oxidative stability of sunflower oil kept in the dark at 70°C.

2. Materials and Methods

2.1 Plant Material and Chemicals

An inquiry is now underway on leaves collected from Sudan's Alfola Agricultural Ar- ea, specifically from *Annona squamosa* and *Vangueria madagascariensis*. specimens recognized by taxonomists from Khartoum, Sudan's National Center for Research on Medicinal and Aromatic Plants (MAPRI). The plants were preserved for the extraction procedure after being dried in various shades and finely ground with an electric mill. Before beginning the extraction process, the seeds were dried out and then ground into a fine powder using an electric mill. All of the solvents ere of analytical grade.

2.2 Preparation of Methanolic Extract

Thirty grams of dried powdered of *Annona squamosa* (ANL) and *Vangueria madagas- cariensis* leaves (VML) were sonicated with eighty percent methanol (Hwasin Tech- nology, Seoul, Korea) to obtain a methanolic extract with a solid to solvent ratio of one to ten (w/v) overnight at room temperature. Whatman No. 1 filter paper was used to filter the methanolic extracts. Next, a rotary evaporator (Buchi, Flawil, Switzerland) was used to extract the solvents. Each extract's yield was calculated and stored at -80°C for later examination.

3. Methods

3.1 Oxidation

Sample (sunflower oil) weighing 50g were placed in 250-ml Erlenmeyer flasks and allowed to oxidize at 70°C in a shaker water bath while kept dark (Kottermann, Hanigsen, Germany). The assessment of oxidative stability was conducted through the periodic analysis of the oil samples' peroxide value (PV) at 0, 2, 4, 8, 16, 24, 32, 48, 56, and 72 hours. Every analysis was carried out twice

3.2 Oxidation of Sunflower Oil Using *Annona Squamosa* and *Vangueria Mada- Gascariensis* Extracts

To determine the oxidative stability of sunflower oil using dried extracts of *Annona squamosa* and *Vangueria madagascariensis*'s leaves 100 g of sunflow- er oil was treated with 100, 250, and 500 mg of leaves methanolic extract. 200 ppm of BHT was used as a comparative level. In an ultrasonic water bath (Bandelin Electronic, Berlin, Germany), the dried extracts and the synthetic antioxidant were combined with a minimum quantity of absolute methanol and added to 100 g of oil. The mixture was then mixed again for ten minutes. The same amount of methanol was used to dissolve the extracts and BHT in order to create a control sample [9]. Samples and a control group were placed in 250- milliliter Erlenmeyer flasks and left to oxidize at 70 degrees Celsius in the dark within a rattling water bath (Kottermann, Germany). As markers for the primary oxidation of the sunflower oil, the peroxide value (PV) and the inhibi- tion of oil oxidation (IO) were employed. Periodically, the PVs were calculated at 0, 6, 12, 24, 32, 48, and 56 hours. IO is equal to 100 percent - (PV increase of samples/PV increase of control) X 100 percent [10].

3.3 Conjugated Diene Test (Bulk Oil System)

Determining conjugated diene (CD) and measuring absorbance at 234 nm were used to assess the oxidative stability of stored sunflower oil (stored at 70°C for 56 hours) as markers for the initial and secondary oxidation of the oil. The CD was calcu- lated at intervals of 0, 6, 12, 24, 36, 48, and 56 hours. Every treatment was adminis- tered three times. Weighing each sample (20–40 mg) into a 25 mL volumetric flask was done. After that, the mixture was fully mixed and isooctane (ACS grade) was add- ed as needed.

Using a Hewlett-Packard 8452 A diode array spectrophotometer, the absorbance was measured at 234 nm. The standard utilized was pure isooctane. The following equation was used to compute the conjugated diene values. According to (11) CD value defined as $A/(C \cdot l)$, where CD stands for conjugated diene, A for absorbance at 234 nm, C for concentration (g/100 ml), and l for route length (cm).

4. Results and Discussion

4.1 Effects of Adding *Annona Squamosa* and *Vangueria Madagascariensis* Ex- tracts on the Stability of Sunflower Oil as Measured by Peroxide Value

The Peroxide Value PV of sunflower oil (control) increased gradually both with and without ASL, ASC, VML, VMC, VMB, and BHT. As shown in Figs. 1, 2, and 3, the control group obtained a maximum PV of 38.5 meqO₂/kg after 56 hours of storage without the addition of extract or BHT. It was discovered that the PVs of sunflower oil containing 100 mg of VML, ASL, ASC, VMC, VMB, or 20 mg BHT were, in that or- der, 12.5, 13, 15, 15.2, 17, and 8 meq O₂ kg⁻¹ (Fig.1).

Upon 56 hours of storage, these samples displayed an IO of 72, 70, 65, 64, 59, and 82%, respectively, in contrast to the control. It follows that every extract that was add- ed had a strong antioxidative action that prevented oil deterioration. The extracts with the highest activity were VML and ASL; however, BHT appeared to have a greater effect. As shown in Figure 2, the PVs of sunflower oil containing 250 mg of VML, ASL, ASC, VMC, VMB, or 20 mg BHT were determined to be, respectively, 11.5, 13, 13, 14, 15, and 8 meq O₂/kg⁻¹. After 56 hours of storage, these samples displayed an IO of 71, 67, 67, 65, and 82%, respectively, as compared to a control. This suggests that every extract used has exceptional antioxidative activity to reduce the deterioration of the oil. The most active was BHT, which was followed by ASL and VML. According to Figure 3, the PVs of sunflower oil containing 500 mg of ASL, VML, ASC, VMC, VMB, or 20 mg of BHT were 7, 7.5, 10, 12.5, 17, and 8 meq O₂ kg⁻¹, in that order. Mariod ,et. al, and Mustafa, evaluated the phenolic compounds of methanolic extracts of *Annona squamosa* and *Vangueria madagascariensis* (leaves, bark, roots, and seedcake) as shown in Fig.1 and HPLC- DAD and the Folin- Ciocalteu method were used to measure the total phenolic content of the ex- tracts under examination. The GAE values for *A. squamosa* roots, *Vangueria madagascariensis* bark, *A. squamosa* bark, and *Vangueria madagascariensis* leaves were 171.5, 170.4, 169.5, and 167.9 g/kg plant extract, respectively [7,8]. Usg the 1,1-diphenyl-2-picrylhydrazyl (DPPH) assay, Mariod, et. al, and Mustafa, results also demonstrated that the extracts of *Vangueria madagascariensis* and *A. squamosa* exhibited antioxidant properties [7,8]. The IC₅₀ values ranged from 7.81 to 62.5 and from 7.81 to 125.0 $\mu\text{g/ml}$, respectively as shown in Table 1.

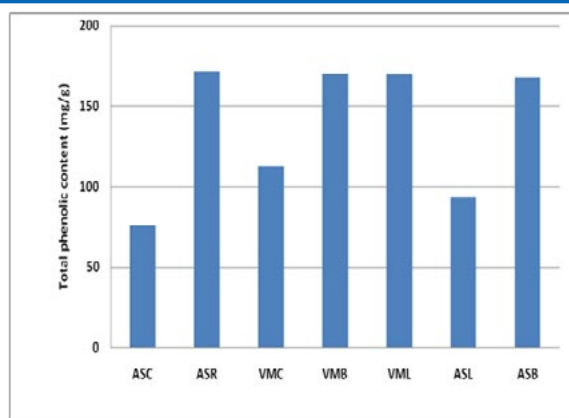


Figure1: The Total Phenolic Content of Methanolic Extracts of *Annona Squamosa*, And *Vangueria Madagascariensis* Bark, Leave, Root and Seedcake

Source:(Mariod.et.al, and Mustafa.et.al.) [7,8].

ASC: *Annona squamosa* seed cake.

ASR: *Annona squamosa* root

VMC: *Vangueria madagascariensis* seed cake.

VMB: *Vangueria madagascariensis* bark.

VML: *Vangueria madagascariensis* leaves.

ASL: *Annona squamosa* leaves.

ASB: *Annona squamosa* bark.

Extract	IC50 $\mu\text{g/ml}$	ORAC (μM of Trolox)
VMB	62.5 \pm 0.5	47.08 \pm 0.23
VMC	31.3 \pm 0.3	44.94 \pm 0.34
VML	7.81 \pm 0.1	72.72 \pm 0.89
ASC	15.6 \pm 0.2	87.95 \pm 0.96
ASB	125.0 \pm 0.4	84.62 \pm 0.31
ASL	7.81 \pm 0.1	29.60 \pm 0.17
ASR	7.81 \pm 0.1	65.20 \pm 0.51
Ascorbic Acid	3.13 \pm 0.1	0.00
Quercetin	0.00	58.97 \pm 0.02

Table1: Results of DPPH (IC50) of *Annona Squamosa*, and *Vangueria madagascariensis* methanolic extracts

*Results are mean \pm SD ($n=3$). Ascorbic Acid used as control for ISC50, while Quercetin was used as control for ORAC and it was found 58.97 \pm 0.02

Source:(Mariod.et.al, and Mustafa.et.al.) [7,8].

VMB: *Vangueria madagascariensis* bark.

VMC: *Vangueria madagascariensis* seed cake.

VML: *Vangueria madagascariensis* leaves.

ASC: *Annona squamosa* seed cake.

ASB: *Vangueria madagascariensis* bark.

ASL: *Annona squamosa* leaves.

ASR: *Annona squamosa* roots.

This result is based on Mariod,et.al, and Mustafa, findings, they discovered that these extracts had good antioxidant activity [7,8]. Additionally, they discovered that the leaf extracts of the two trees had high levels of flavonoids, polyphenols and antioxidant activity. After 56 hours of storage, these samples' respective IOs were 83, 82, 75, 70, 57, and 82% as compared to the control. We conclude that 0.5% concentrations of *Vangueria madagascariensis* leaf extract effectively stabilized sunflower oil during 70°C storage. The

ability of antioxidant components found in *Annona squamosa* and *Vangueria madagascariensis* leaves to protect oil from oxidation is significantly impacted by this. Based on this analysis, the potential value of extracts from *Annona squamosa* leaves, and *Vangueria madagascariensis* leaves was shown to be higher than that of BHT. These extracts may have use as prophylactic antioxidant agents and for enhancing the nutritional value of foods in the future.

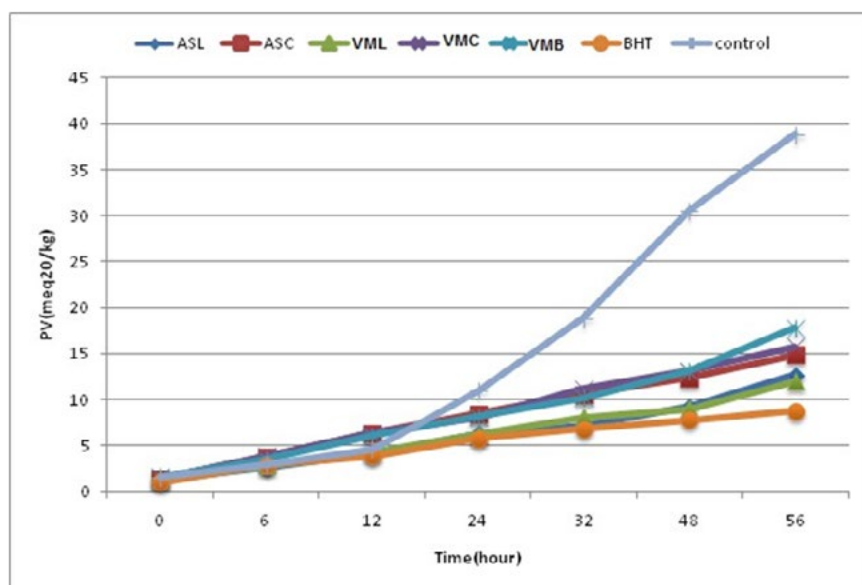


Figure 2: Oxidation of Sunflower Oil Treated with *Annona Squamosa*, And *Vangueria Madagascariensis* Ex- Tracts 100mg/100g Oil During Storage At 70°C

ASL: *Annona squamosa* leaves

ASC: *Annona squamosa* seed cake.

VML: *Vangueria madagascariensis* leaves.

VMC: *Vangueria madagascariensis* seed cake.

VMB: *Vangueria madagascariensis* bark.

ASB: *Vangueria madagascariensis* bark.

BHT: Butylated hydroxyl toluene

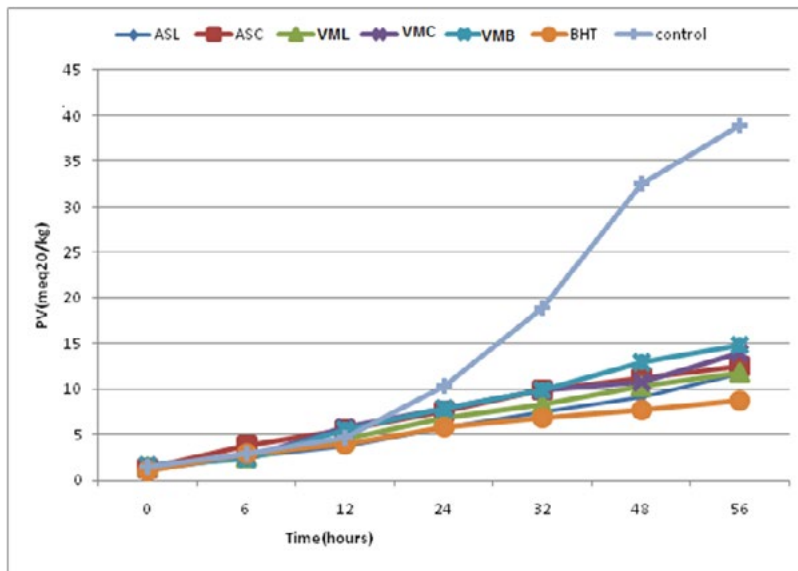


Figure 3: Oxidation of Sunflower Oil Treated with *Annona Squamosa*, and *Vangueria Madagascariensis* Ex- Tracts 250mg/100g Oil During Storage At 70°Cp

ASL: *Annona squamosa* leaves

ASC: *Annona squamosa* seed cake.

VML: *Vangueria madagascariensis* leaves.

VMC: *Vangueria madagascariensis* seed cake.

VMB: *Vangueria madagascariensis* bark.

BHT: Butylated hydroxyl toluene

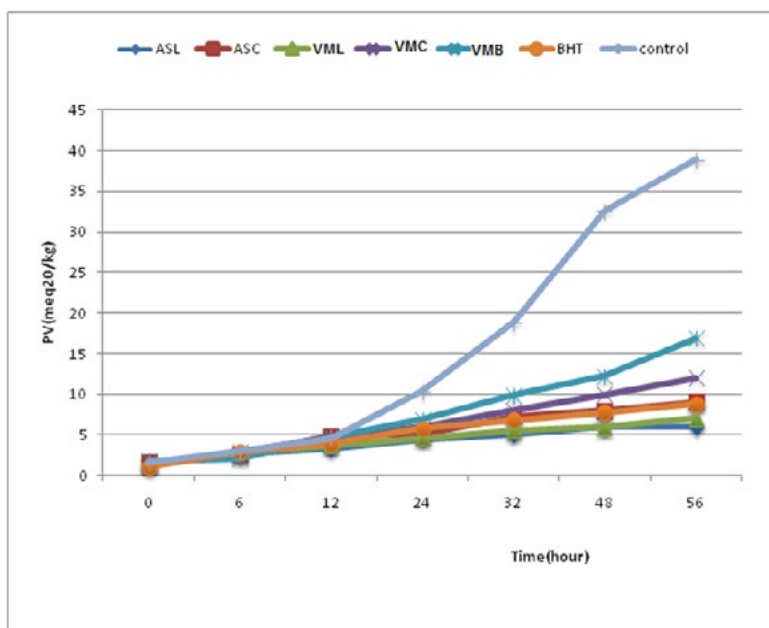


Figure 4: Oxidation of Sunflower Oil Treated with *Annona Squamosa*, and *Vangueria Madagascariensis* Ex- Tracts 500mg/100g Oil During Storage At 70°C

ASL: *Annona squamosa* leaves

ASC: *Annona squamosa* seed cake.

VML: *Vangueria madagascariensis* leaves.

VMC: *Vangueria madagascariensis* seed cake.

VMB: *Vangueria madagascariensis* bark.

BHT: Butylated hydroxyl toluene

4.2 The Effects of Adding Various Extracts from *Annona Squamosa* and *Vangueria Madagascariensis* (ASL, VML, ASC, VMC, VMB) on the Stability of Sunflower Oil as Measured by Conjugated Dienes

The degree of oxidative deterioration has been assessed using a number of techniques that are connected to the determination of the concentration of primary or secondary oxidation products. Specific absorptivity in the UV region at 232 and 270 nm is one of the most widely utilized. Conjugated dienes (CDs) and conjugated trienes (CTs) are measured by the specific absorptivity at 232 and 270 nm [12]. Polyunsaturated Acids (PUFA) undergo a double bond shift that results in the formation of conjugated dienes as intermediates. UV absorbance at 232-234 nm can be used to quantify these substances [4].

The ability of methanolic extracts from *Annona squamosa* and *Vangueria madagascariensis* to suppress the oxidation of edible sunflower oil at 70°C at 234 nm (while monitoring conjugated dienes) was studied at 100, 250, and 500 mg (Figs. 4, 5 and 6). According to Figure 6, all of the extracts at 500 mg/100 g of oil seemed to be potent oxidation inhibitors of sunflower oil. The leaves of *Annona squamosa* and *Vangueria madagascariensis* were the most active extracts; at 20 mg/kg, these extracts' antioxidant activity seemed comparable to that of BHT. The antioxidant activity of the other three extracts—ASC, VMC, and VMB—was less than that of BHT.

According to the above-mentioned study results, extracts from *Annona squamosa* and *Vangueria madagascariensis* leaves, seedcakes, and bark appear to be a good source of naturally occurring antioxidants that can be used to prevent edible sunflower oil oxidation. These findings were consistent with those of who investigated various extracts from various plants sources to be used as edible oil oxidation inhibitors[12]. Figure 6 shows the conjugated diene of sunflower oil on hours 0 and 56 of storage, incorporating extracts of *Annona squamosa* and *Vangueria madagascariensis* leaves at 500 mg and BHT at 200 ppm/100 g (because it was better than 250 mg). In this investigation, the antioxidant-rich sunflower oil demonstrated noticeably decreased CD production than the antioxidant-free oil (control). After 56 hours of storage, the conjugated diene values of sunflower oil containing extracts of *Annona squamosa* and *Vangueria madagascariensis* increased by two to three times, whereas the control samples exhibited an increase of five times or more. The extracts and controls that were employed showed a decrease in oxidation inhibitory activity in the following order: BHT < VML < ASL < ASC < VMC < VMB < control. This outcome stems from studies conducted by (6 and 7), which found that extracts from the leaves, seed cakes, and bark of *Annona squamosa* and *Vangueria madagascariensis* had good antioxidant activity. They also found that there was strong flavonoid and polyphenol content as well as antioxidant activity in the leaf extracts of the two plants.

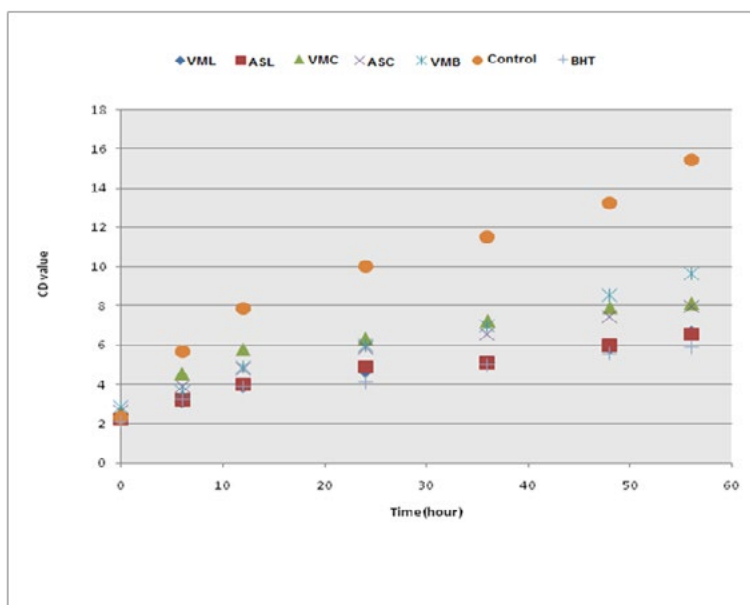


Figure 5: Conjugated Diene Values of Annona Squamosa and Vangueria Madagascariensis Leaves, Seed Cakes, and Vangueria Madagascariensis Bark Phenolic Extracts At 100 Mg/100 G and Bht at 200 Ppm in Sunflower Oil

VML: Vangueria madagascariensis leaves.
 ASL: Annona squamosa leaves
 VMC: Vangueria madagascariensis seed cake.
 ASC: Annona squamosa seed cake.
 VMB: Vangueria madagascariensis bark.
 BHT: Butylated hydroxyl toluene

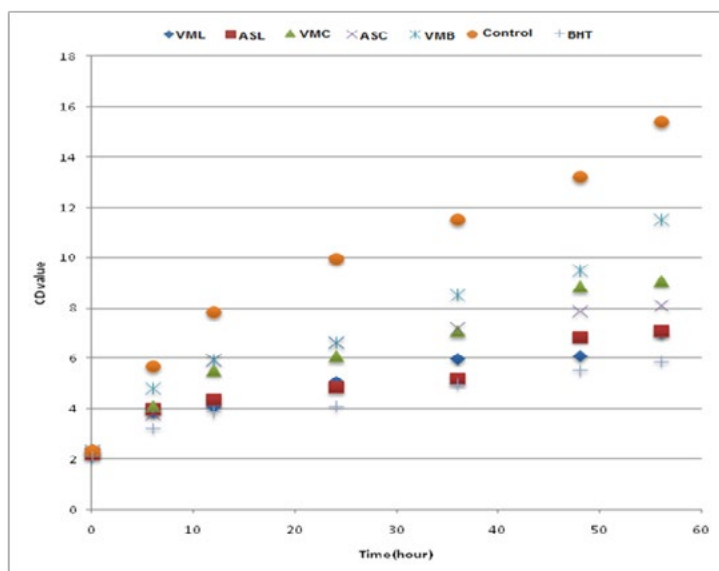


Figure 6: Conjugated Diene Values of Annona Squamosa and Vangueria Madagascariensis Leaves, Seed Cakes, and Vangueria Madagascariensis Bark Phenolic Extracts at 250mg/100 g and BHT at 200 ppm in Sunflower Oil

VML: Vangueria madagascariensis leaves.
 ASL: Annona squamosa leaves
 VMC: Vangueria madagascariensis seed cake.
 ASC: Annona squamosa seed cake.
 VMB: Vangueria madagascariensis bark.
 BHT: Butylated hydroxyl toluene

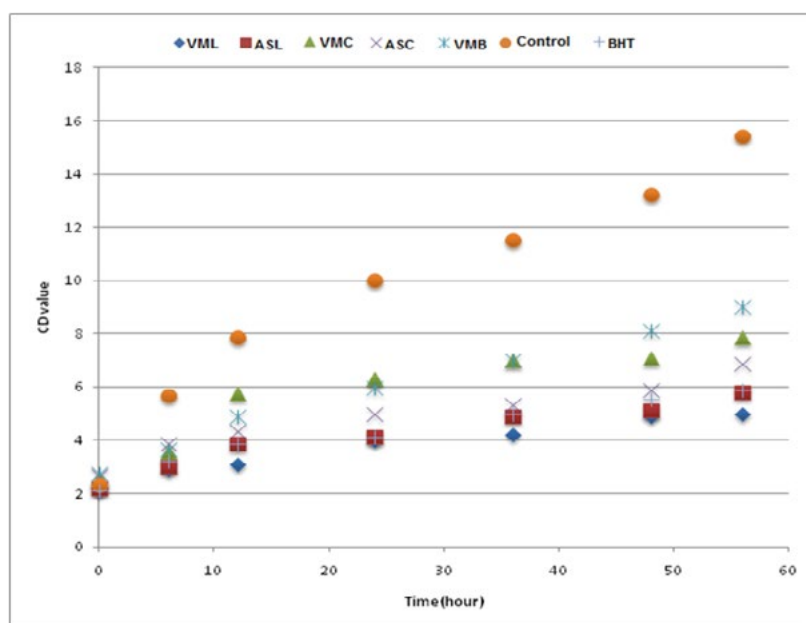


Figure 7: Conjugated Diene Values of *Annona Squamosa* and *Vangueria Madagascariensis* Leaves, Seed Cakes, and *Vangueria Madagascariensis* Bark Phenolic Extracts at 500mg/100 g and BHT at 200 ppm in Sunflower Oil

VML: *Vangueria madagascariensis* leaves.

ASL: *Annona squamosa* leaves

VMC: *Vangueria madagascariensis* seed cake.

ASC: *Annona squamosa* seed cake.

VMB: *Vangueria madagascariensis* bark.

BHT: Butylated hydroxyl toluene

5. Conclusions

We may conclude that extracts from *Annona squamosa* and *Vangueria madagascariensis* leaves, at 0.5% (w/w) concentrations, were successful in stabilizing sunflower oil when it was stored at 70 °C. This has a significant effect on extending research into the antioxidant components found in these plant extracts and their application in preventing oil oxidation.

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